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Subject : Chemistry

Class : B.Sc. 2nd Year Semester IV

Paper II : PHYSICAL CHEMISTRY

Unit I

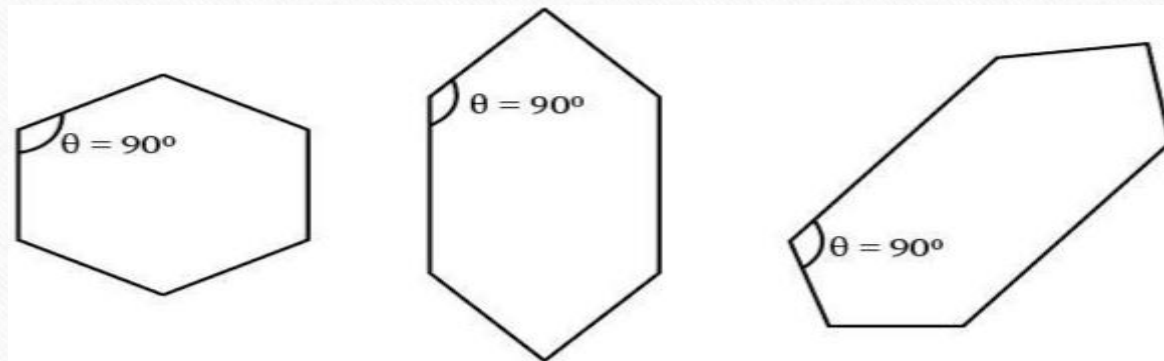
TOPIC : Law of Crystallography

Crystallography

- It is a branch of science which deals with geometrical properties and structure of Crystal and crystalline substances.
- Crystallography is based on three fundamental laws. They are-
 1. Law of constancy of interfacial angle
 2. Law of rationality of indices
 3. Law of symmetry

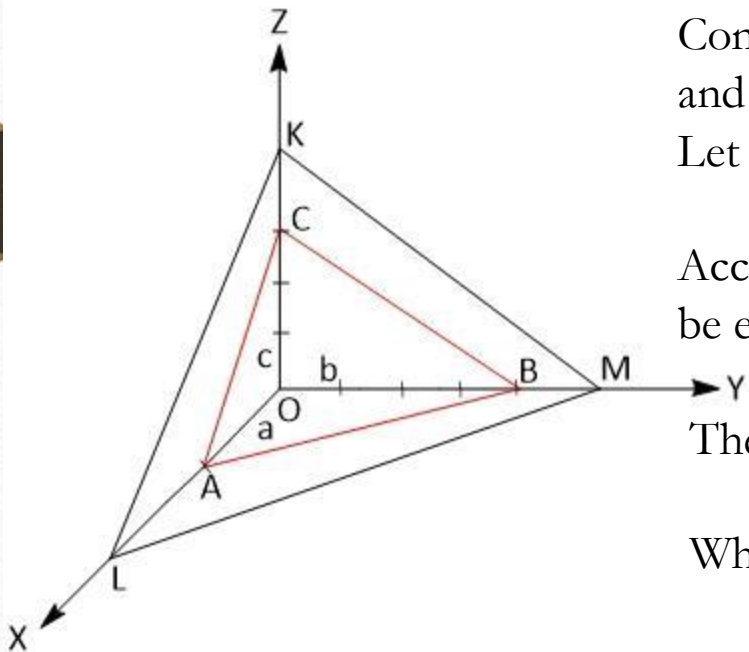
Law of constancy of interfacial angle

- These law was given by steno.
- *“The crystals of same substance may have different shapes depending upon sizes of faces but the angles between the corresponding faces remain constant”.*
- It can be well understood from the diagrams as given below



Law of rationality of indices

- These law was given by **Hauy**.
- “*All the phases of a crystal lattice cut the given Axis at different positions which are the integral multiples of unit plane.*”



Consider a 3D coordinate system with x, y, & z axes. A plane ABC is unit plane and KLM is any face parallel to ABC plane.

Let the distances are $d(OA) = a$ $d(OB) = b$ $d(OC) = c$

According to law of rationality of indices the intercepts made by plane KLM can be expressed as-

$$d(OL) = la \quad d(OM) = mb \quad d(OK) = nc$$

The ratios are given by-

$$d(OP) / d(OA) = l \quad d(OM) / d(OB) = m \quad d(OK) / d(OC) = n$$

Where, l, m, n are simple integral wholes numbers or fractions of whole numbers.

Law of symmetry

- “All the crystals of same substance possess same elements of symmetry”.
- It can be well understood from the elements of symmetry

• **Elements of Symmetry:** The crystal has 3 types of elements of symmetry.

1. **Plane of symmetry:** It is defined as “An imaginary plane by which, if a crystal is divided into two parts one becomes the mirror image of each other”.

In cubic crystal system, it have total **9** planes of symmetry as shown in diagram

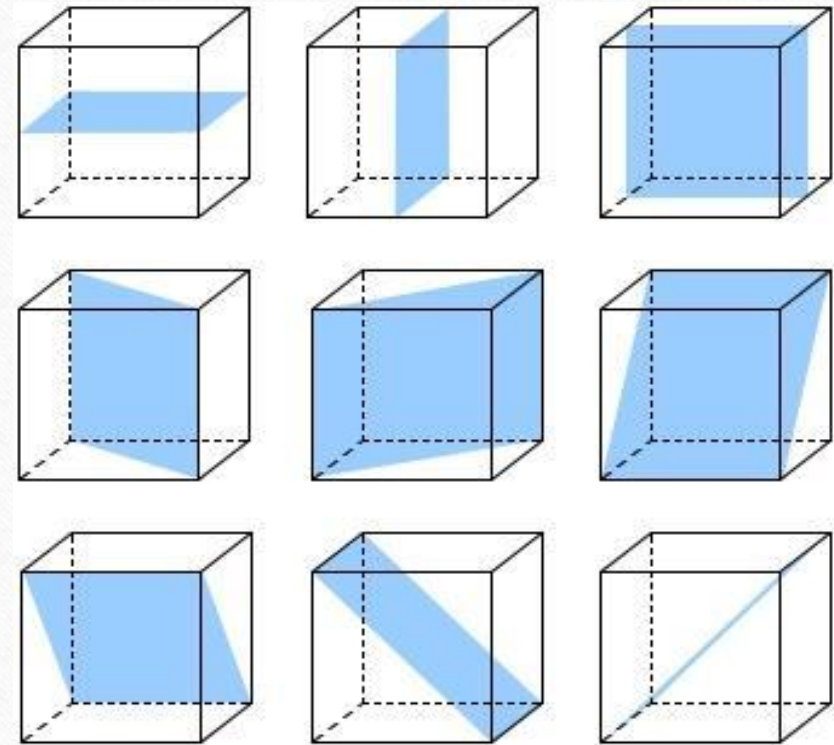
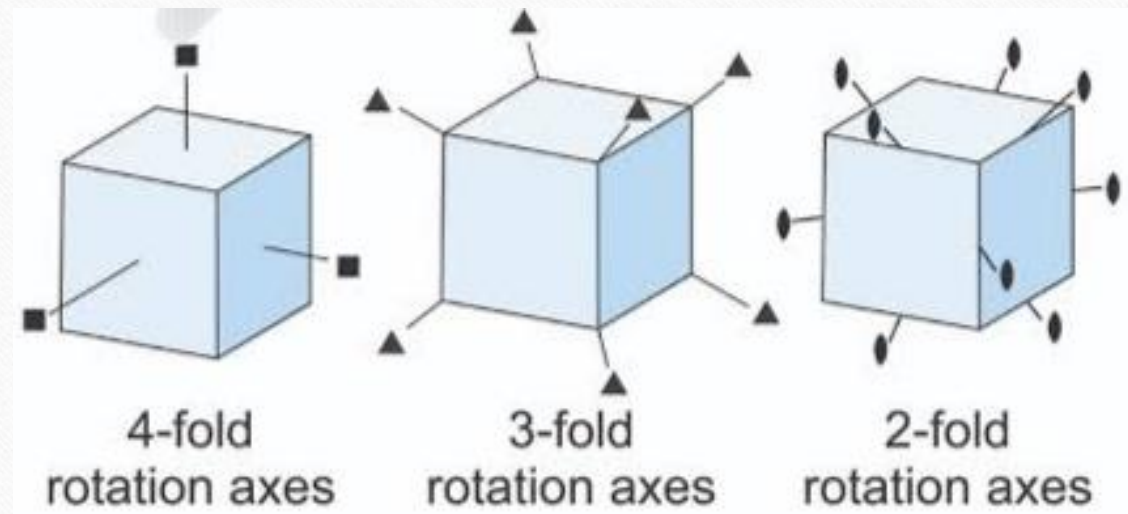


Fig. Plane of symmetry in cube

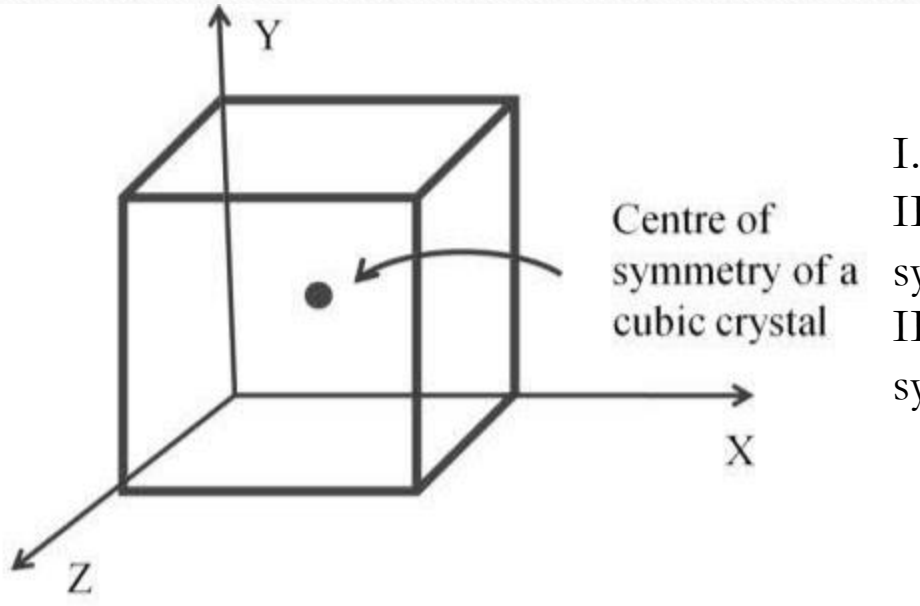
2. **Axis of symmetry:** It is defined as, “An imaginary line through the Crystal about which is the Crystal is rotated through 360° it presents the same appearance more than once during the course of complete rotation”.



In cubic crystal system, it shows total **13** planes of symmetry as shown in figure.

Fig. Axis of symmetry in cube

3. Centre of symmetry: it is defined as “Centre or point of symmetry of a crystal is point within the Crystal such that any line drawn through it will intersect the surface of the Crystal at equal distance in both direction.



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- I. In cubic t system it has **1** center of symmetry.
 - II. Sum of all the number of a planes access and points of symmetry are called as **element of symmetry**.
 - III. For cubic crystal system the there are total **23** elements of symmetry.

Definition

- **Space lattice:** Regular arrangement of the constituent particles of a crystalline solid in three dimensional space is called as space lattice.
- **Lattice point:** Corresponding to each particle, there is a point in the space lattice which is called as lattice point.
- **Unit cell:** The smallest repeating unit in space lattice which when repeated over and over produces the complete space lattice is called as unit cell.